



SAFETY DATA SHEET

Hydrofluoric Acid 10:1

SECTION 1: Identification

1.1 GHS Product identifier

Product name Hydrofluoric Acid 10:1

1.2 Other means of identification

Hydrofluoric acid solution; Fluohydric acid; Fluoric acid

1.3 Recommended use of the chemical and restrictions on use

For Laboratory, Research or Manufacturing Use.

1.4 Supplier's details

Name High Purity Products
Address 14546 N. Lombard Street
Portland OR 97203
United States of America

Telephone 503-227-1616
email help.desk@highpp.com

1.5 Emergency phone number

CHEMTREC: 1-800-424-9300

SECTION 2: Hazard identification

2.1 Classification of the substance or mixture

GHS classification in accordance with: OSHA (29 CFR 1910.1200)

- Corrosive to metals, Cat. 1
- Skin corrosion/irritation, Cat. 1A
- Eye damage/irritation, Cat. 1
- Acute toxicity, oral, Cat. 4
- Acute toxicity, dermal, Cat. 3
- Acute toxicity, inhalation, Cat. 4

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2.2 GHS label elements, including precautionary statements

Pictogram



Signal word

Danger

Hazard statement(s)

H290 May be corrosive to metals
H302 Harmful if swallowed
H311 Toxic in contact with skin
H314 Causes severe skin burns and eye damage
H318 Causes serious eye damage
H332 Harmful if inhaled

Precautionary statement(s)

P234 Keep only in original container.
P260 Do not breathe vapors.
P262 Do not get in eyes, on skin, or on clothing.
P264 Wash hands thoroughly after handling.
P280 Wear protective gloves/protective clothing/eye protection/face protection.
P284 In case of inadequate ventilation wear respiratory protection.
P301+P310 IF SWALLOWED: Immediately call POISON CENTER
P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.
P302+P361+P354 IF ON SKIN: Take off Immediately all contaminated clothing. Immediately rinse with water for several minutes and apply Use calcium gluconate gel.
P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses if present and easy to do. Continue rinsing.
P310 Immediately call a POISON CENTER
P361+P364 Take off immediately all contaminated clothing and wash it before reuse.
P390 Absorb spillage to prevent material damage.
P403+P233 Store in a well-ventilated place. Keep container tightly closed.

SECTION 3: Composition/information on ingredients

3.1 Mixture

Component	CAS #	Concentration
Hydrofluoric acid	7664-39-3	4.5 – 5.5 % (weight)
Water	7732-18-5	94.5 – 95.5 % (weight)

SECTION 4: First-aid measures

4.1 Description of necessary first-aid measures

General advice Consult a physician. Show this safety data sheet to the doctor in attendance. Move out of dangerous area.

If inhaled If breathed in, move person into fresh air. If not breathing, give artificial respiration. Consult a physician.

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In case of skin contact	Take off contaminated clothing and shoes immediately. Wash off with soap and plenty of water. If available, apply calcium gluconate gel (2.5%) into burn area continuously for 15 minutes or until pain relief. For a larger area, use iced Benzalkonium Chloride 0.13% soaks until pain has resolved at least 30-40 minutes. If calcium gluconate gel or Benzalkonium Chloride is not available, continue to wash exposed areas with water until patient is seen by a physician and is taken to a hospital.
In case of eye contact	Rinse thoroughly with plenty of water for at least 15 minutes and consult a physician. Continue rinsing eyes during transport to hospital. Keep the eyelids apart and away from the eyeballs during irrigation. Do not use oily drops or ointment or HF skin burn treatments on the eyes.
If swallowed	Call a physician or poison control center immediately. Rinse mouth. Do NOT induce vomiting. If vomiting occurs, keep head low so that stomach content doesn't get into the lungs.

4.2 Most important symptoms/effects, acute and delayed

Causes severe skin and eye burns. Causes digestive tract burns. Mist or vapor extremely irritating to eyes and respiratory tract.

4.3 Indication of immediate medical attention and special treatment needed, if necessary

Injection of 5% calcium gluconate is indicated as the primary medical treatment for large burns. If benzalkonium chloride soaks or calcium gluconate gel, do not provide significant relief of pain within 30 to 40 minutes, injection of calcium gluconate solution is indicated. For burns of large skin areas (>15%), for ingestion and for significant inhalation exposure, severe systemic effects may occur. Monitor and correct for hypocalcemia, cardiac arrhythmias, hypomagnesemia and hyperkalemia. Calcium supplements are essential for emergency response to large exposures. Symptoms may be delayed.

SECTION 5: Fire-fighting measures

5.1 Suitable extinguishing media

Use extinguishing media appropriate for surrounding fire.

5.2 Specific hazards arising from the chemical

During fire, gases hazardous to health may be formed. Product is highly caustic. Product is acidic. Wear appropriate protective gear if spilled during firefighting.

5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary. Emits toxic fumes (hydrogen fluoride) under fire conditions.

SECTION 6: Accidental release measures

6.1 Personal precautions, protective equipment and emergency procedures

Wear respiratory protection. Avoid breathing vapors, mist or gas. Ensure adequate ventilation. Evacuate personnel to safe areas.

6.2 Environmental precautions

Prevent further leakage or spillage if safe to do so. Do not let product enter drains.

6.3 Methods and materials for containment and cleaning up

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Neutralize with lime or soda ash. Neutralize spill area and washings with dilute acetic acid. Absorb spill with vermiculite or other inert material, then place in a container for chemical waste. Dike far ahead of larger spill for later recovery and disposal.

SECTION 7: Handling and storage

7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid inhalation of vapor or mist. For precautions see section 2.2.

7.2 Conditions for safe storage, including any incompatibilities

Keep container tightly closed in a dry and well-ventilated place. Containers which are opened must be carefully resealed and kept upright to prevent leakage.

SECTION 8: Exposure controls/personal protection

8.1 Control parameters

Hydrogen fluoride (7664-39-3)

TWA 3 ppm (2.5 mg/m³) (NIOSH)

C 6 ppm (5 mg/m³) [15-minute] (NIOSH)

8.2 Appropriate engineering controls

Use ventilation adequate to keep exposures (airborne levels of vapor) below recommended exposure limits.

8.3 Individual protection measures, such as personal protective equipment (PPE)

Pictograms



Eye/face protection

Tightly fitting safety goggles. Face shield (8-inch minimum). Use equipment for eye protection tested and approved under appropriate government standards

Skin protection

Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands.

Respiratory protection

When workers are facing concentrations above the exposure limit they must use appropriate certified respirators.

SECTION 9: Physical and chemical properties and safety characteristics

Physical state	Liquid
Appearance	Clear Liquid
Color	Colorless
Odor	Mild pungent
Odor threshold	0.5 - 3 ppm
pH	1.4 ~
Melting point/freezing point	No data available.
Boiling point or initial boiling point and boiling range	No data available.

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Flash point	Not Flammable
Evaporation rate	No data available.
Flammability	Not Flammable
Vapor pressure	No data available.
Relative vapor density	No data available.
Density and/or relative density	1.013 g/cm ³
Solubility	Miscible
Auto-ignition temperature	Not Flammable
Decomposition temperature	No data available.
Kinematic viscosity	No data available.

SECTION 10: Stability and reactivity

10.1 Reactivity

Reacts violently with strong alkaline substances.

10.2 Chemical stability

Stable under recommended storage conditions.

10.3 Possibility of hazardous reactions

Metals, Cyanides, Sulfides, Bases, Fluorine

10.4 Conditions to avoid

Heat, sparks, flames. Contact with incompatible materials.

10.5 Incompatible materials

Strong oxidizing agents. Acids. Strong bases. Ammonia. Organic compounds. Glass. Cyanides. Fluorine. Metals. May attack some plastics, rubber and coatings

10.6 Hazardous decomposition products

Gaseous hydrogen fluoride

SECTION 11: Toxicological information

Information on toxicological effects

Acute toxicity

LD50 Inhalation - guinea pig – 4327 ppm/15 M

LD50 Inhalation - mouse – 342 ppm/ 1 H

Skin corrosion/irritation

Causes severe burns by all exposure routes

Serious eye damage/irritation

Burns, pain, watering eyes

Inhalation

Burning, choking, coughing, wheezing, laryngitis, shortness of breath, headache or nausea.

Germ cell mutagenicity

May cause genetic effects based on animal data.

Carcinogenicity

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Not listed by IARC NTP ACGIH OSHA

Reproductive toxicity

Reproductive Effects – Inhalation - Rat – 4980 µg/m³/ 4H - Effects on embryo or fetus: Fetal death

STOT-single exposure

Blood, Cardiovascular system, Respiratory system, Skin

STOT-repeated exposure

Bones, Endocrine system, Teeth

SECTION 12: Ecological information

Toxicity

Aquatic fish; EC50 (48 hours): 270 mg/l

Leuciscus idus; LC50 (48 hours): 660mg/L

Persistence and degradability

Hydrogen fluoride is removed from air by wet deposition as fluoride salts with an atmospheric lifetime of 1-5 days. The substance is harmful to aquatic organisms.

Bioaccumulative potential

Inherently biodegradable.

Mobility in soil

Will likely be mobile in the environment due to its water solubility.

SECTION 13: Disposal considerations

Product disposal

Waste material must be disposed of in accordance with the national and local regulations.

Leave chemicals in original containers. No mixing with other waste. Handle uncleaned containers like the product itself.

EPA Waste Code: **U134** (C,T)

SECTION 14: Transport information

DOT (US)

UN Number: UN1790

Class: 8 (6.1)

Packing Group: II

Proper Shipping Name: Hydrofluoric acid, (with < 6% Strength)

Reportable quantity (RQ): 2,000 lbs

IMDG

UN Number: UN1790

Class: 8 (6.1)

Packing Group: II

Proper Shipping Name: Hydrofluoric acid, (with < 6% Strength)

IATA

UN Number: UN1790

Class: 8 (6.1)

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Packing Group: II
Proper Shipping Name: Hydrofluoric acid, (with < 6% Strength)

SECTION 15: Regulatory information

15.1 Safety, health and environmental regulations specific for the product in question

Massachusetts Right To Know Components

Chemical name: Hydrofluoric acid
CAS number: 7664-39-3

New Jersey Right To Know Components

Common name: HYDROGEN FLUORIDE
CAS number: 7664-39-3

Pennsylvania Right To Know Components

Chemical name: Hydrofluoric acid
CAS number: 7664-39-3

Canadian Domestic Substances List (DSL)

Chemical name: Hydrofluoric acid
CAS: 7664-39-3

EPCRA Section 302 (EHS) TPQ Extremely Hazardous Substances

Hydrofluoric acid: 100 lbs

EPCRA Section 304 EHS RQ Reportable Quantities

Hydrofluoric acid: 100 lbs

CERCLA RQ Hazardous Substances

Hydrofluoric acid: 100 lbs

EPCRA Section 313 Toxic chemicals

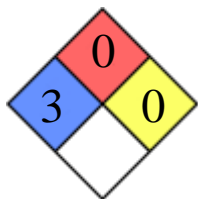
Hydrofluoric acid

HMIS Rating

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HEALTH	*3
FLAMMABILITY	0
PHYSICAL HAZARD	0
PERSONAL PROTECTION	

NFPA Rating

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SECTION 16: Other information

16.1 Further information/disclaimer

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